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### Acids Bases And Salt

Workbook Acids, Bases and Salts.

Worksheet 1. 1) What is the color of.

(a) Phenolphthalein in Acetic Acid.

(b) Methyl Orange in NaOH.

(c) Blue litmus in Magnesium Hydroxide.

(d) Red Litmus in Ammonium Hydroxide.

2) You are provided with three test tubes. One contains water; the other two contain an acid.

acids bases and salts worksheet 1 | Acid | Sodium Chloride ... Acid Base and Salt

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## Key

Science Chapter 5: Acid Base and Salt Acids, Bases and Salts Hebden - Unit 4 (page 109-182) We will cover the following topics: 1.

Amphiprotic Substance 2.

Amphoteric Compounds CHEM 0012

Lecture Notes 17 Acids, Bases and Salts Hebden - Unit 4 (page

109-182) 1. Amphiprotic Substance

A substance that can act either as a proton acceptor or a proton

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... a) Limewater is the common

name for a diluted solution of

calcium hydroxide c)Range 1-3

d)Strong Base is NaOH and Weak

Base is  $\text{NH}_4\text{OH}$  e) A hydrated salt

is copper sulphate crystals ( $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ ).

f) A universal indicator is a mixed indicator of organic

Key

chemicals which not only shows whether the given solution is acidic or basic, but also shows the approximate pH values by giving a wide ... Class 10 Science Acid base and Salts Practice

worksheet Bronsted-Lowry base = proton acceptor ( $H^+$  = proton)

Bronsted-Lowry acid-base reaction = reaction involving the transfer of a proton e.g.  $HCl + NaOH \rightarrow NaCl + H_2O$   $H^+$  donates  $H^+$   $\therefore$  acid  $OH^-$  accepts  $H^+$   $\therefore$  base e.g.  $HNO_3 + NH_3 \rightarrow NH_4^+ + NO_3^-$   $H^+$  donates  $H^+$   $\therefore$  acid  $NH_3$  accepts  $H^+$   $\therefore$  base

TASK 1 - Bronsted-Lowry acids & bases © www.CHEMSHEETS.co.uk

14-Jul-12 Chemsheets A2 009 1 1.

acids 2. citric acid 3. Base 4.

calcium sulphate hemihydrate,  $CaSO_4 \cdot \frac{1}{2}H_2O$ . 5. yellow colour, pink colour 6. Hydrochloric Acid

Key

(HCl) 7. Acetic Acid ( $\text{CH}_3\text{COOH}$ ) 8. Sodium Hydroxide ( $\text{NaOH}$ ) 9. Ammonium Hydroxide ( $\text{NH}_4\text{OH}$ ) 10. sodium carbonate,  $\text{Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O}$ , permanent hardness 11. PH 12. Acid 13. Base 14. Neutral 15. decreases 16 ... CBSE Class 10 - Science - CH2 - Acids, Bases and Salts ... Acid and Base Worksheet - Answers. 1) Using your knowledge of the Brønsted-Lowry theory of acids and bases, write equations for the following acid-base reactions and indicate each conjugate acid-base pair: a)  $\text{HNO}_3 + \text{OH}^- \rightarrow \text{H}_2\text{O} + \text{NO}_3^-$ .  $\text{HNO}_3$  and  $\text{NO}_3^-$  make one pair  $\text{OH}^-$  and  $\text{H}_2\text{O}$  make the other. b)  $\text{CH}_3\text{NH}_2 + \text{H}_2\text{O} \rightarrow \text{CH}_3\text{NH}_3^+ + \text{OH}^-$  Acid and Base Worksheet - Answers Some of the worksheets displayed are , Acids bases and solutions answer key, Acids bases

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Teacher Worksheets Compounds

formed from combinations of acids  
and bases taste salty and are

referred to in chemistry as salts.

Sweet compounds have

characteristics of both acids and



## Key

bases in the same molecule. We will explore the relationship between molecular structure and acids–bases, and consider water solutions of acids and bases. Introduction to Acids and Bases (Worksheet) - Chemistry ... Acids and bases are chemical opposites of each other. Though it is not a good idea to taste chemicals, acids have a sour taste and bases taste bitter. When they are in solution with water, acids feel rough and bases feel slippery. Some acids and bases are present in foods and in household items. These are relatively safe to handle. Natural Sciences Grade 7 - Grade 7-9 Workbooks Ch19 Acids Bases And Salts Key. Ch19 Acids Bases And Salts Key - Displaying top 8 worksheets found for this

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... CLASS WORKSHEET 5.1 - ACIDS, BASES AND SALTS 1) Review of

Cations, Anions and Molecules

CATIONS ANIONS MOLECULES

ammonium  $\text{NH}_4^+$  + aluminium  $\text{Al}^{3+}$

barium  $\text{Ba}^{2+}$  calcium  $\text{Ca}^{2+}$  ...

Classifying substances as acids,

bases and salts ACIDS BASES SALTS

$\text{HNO}_3$   $\text{Na}_2\text{O}$   $\text{NH}_4\text{NO}_3$   $\text{HCl}$

$\text{Ca}(\text{OH})_2$   $\text{SrSO}_4$   $\text{H}_2\text{SO}_4$   $\text{K}_2\text{CO}_3$

$(\text{NH}_4)_2\text{SO}_4$   $\text{MgO}$   $\text{K}_2\text{SO}_4$   $\text{MgCO}_3$

$\text{Ca}(\text{NO}_3)_2$  ... CLASS WORKSHEET

5.1 - ACIDS, BASES AND

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