

Chemical Quantities Chapter Test

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- The basic unit that is used to determine the amount of a chemical substance is called a mole
- A mole(mol) of a substance is equivalent to 6.02×10^{23} particles of that substance
- The mole was founded by a scientist named Avagadro, and he decided to use the Chapter 10 Chemical Quantities Quiz Answer Key Description Of

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Answers atomic mass. the number of representative particles in a mole of a substrate. Avogadro's number. the percent by mass of each element in a compound. percent composition. the mass of a mole of a substrate. molar mass. the formula that gives the lowest whole-number ratio of the elements in a compound. empirical data. Chapter 10 Chemical Quantities Flashcards | Quizlet chemistry chapter 10 chemical quantities study flashcards learn write spell test play match gravity created by destiny yarrington terms in this set 23 602 x 10²³ the number of particles in exactly one ... chapter 10 chemical quantities guided reading answers pearson Golden Education World Book Chapter 10 Chemical Quantities Guided Reading

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