

# Chemistry Chapter 13 Answers

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Chemistry Chapter 13 Answers Answers to Chapter 13 Study Questions. 1. True: b), c) and d) 2. Reaction = 2 x reverse of original reaction. So  $K_2 = (1/K_1)^2 = (1/0.447)^2 = 5.00$ . 3.  $K = [\text{NH}_3]^2 / [\text{N}_2] \times [\text{H}_2]^3$ ; more product made. Increasing the pressure increases the concentration; counteract this by shifting to the side with fewer moles. Answers to Chapter 13 Study Questions - Chemistry LibreTexts Mastering Chemistry Chapter 13 Homework Mastering chemistry chapter 13 answers. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. madievens4. Terms in this set (28) Determine the volume of 0.170 M

NaOH solution required to neutralize each sample of hydrochloric acid. The neutralization reaction is:  $\text{NaOH(aq)} + \text{HCl(aq)} \rightarrow \text{H}_2\text{O(l)} + \text{NaCl(aq)}$  20 mL of a 0.170 M HCl solution. Mastering Chemistry Chapter 13

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has reached equilibrium, no further changes in the reactant and product concentrations occur; the reactions continue to occur, but at equivalent rates.

5. 13.1 Chemical Equilibria - Chemistry Mastering  
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derived by replacing one or more hydrogen atoms of  
ammonia molecule by alkyl/aryl group (s). In nature,  
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